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NEWS EXPRESS OCTOBER 29 CURRENT WINDOWS VERSION IS V7.01A, CURRENT MACINTOSH VERSION IS V6.0c(ENG) AND V6.0Jc(JP), AND CURRENT DISCOVER FILE IS DATED 11 AUGUST 2004 NEWS HOURS STN Operating Hours Plus Help Desk Availability NEWS INTER General Internet Information Welcome Banner and News Items NEWS LOGIN Direct Dial and Telecommunication Network Access to STN

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* * * * * * * * * * * * * * STN Columbus

FILE 'HOME' ENTERED AT 15:31:36 ON 15 DEC 2004

=> file caplus wpids COST IN U.S. DOLLARS

NEWS PHONE

SINCE FILE TOTAL ENTRY SESSION 0.21 0.21 FULL ESTIMATED COST

FILE 'CAPLUS' ENTERED AT 15:31:48 ON 15 DEC 2004 USE IS SUBJECT TO THE TERMS OF YOUR STN CUSTOMER AGREEMENT. PLEASE SEE "HELP USAGETERMS" FOR DETAILS. COPYRIGHT (C) 2004 AMERICAN CHEMICAL SOCIETY (ACS)

FILE 'WPIDS' ENTERED AT 15:31:48 ON 15 DEC 2004 COPYRIGHT (C) 2004 THE THOMSON CORPORATION

=> s alkylphenol amine

L1110 ALKYLPHENOL AMINE

=> s 11 and (emulsion or microemulsion or emulsified)

3 L1 AND (EMULSION OR MICROEMULSION OR EMULSIFIED) L2

=> d 12 all

ANSWER 1 OF 3 CAPLUS COPYRIGHT 2004 ACS on STN L2

Citing Full References Text

1983:91204 CAPLUS AN

98:91204 DN

h

Entered STN: 12 May 1984 ED

eb c g cg b cg

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Vyskocil, Ivan; Hrdina, Daniel; Paulovic, Milan
IN
PA
SO
    Czech., 2 pp.
    CODEN: CZXXA9
DT
    Patent
LA
    Slovak
    C09D003-40
IC
    42-11 (Coatings, Inks, and Related Products)
CC
FAN.CNT 1
                                          APPLICATION NO.
                                                               DATE
                      KIND
                               DATE
    PATENT NO.
                        ____
    CS 191662
                        В
                                          CS 1977-2300
                                                               19770407
                               19790731
PΙ
                       Α
                               19770407
PRAI CS 1977-2300
CLASS
               CLASS PATENT FAMILY CLASSIFICATION CODES
PATENT NO.
 IC
                       C09D003-40
CS 191662
    Slipperiness of floors is prevented by application of an aq. emulsion
    contg. 3-10% tall-oil rosin and 5-10% nonionic emulsifiers of HLB value
    8-18 based on ethoxylated fatty acids, alcs., alkylphenols, amines, or
    esters. A typical emulsion comprised tall-oil rosin 70, Slovasol EL
     (castor oil oxyethylated with 20 mol oxirane) 15, and Slovasol A
     [9004-96-0] (oleic acid oxyethylated with 6 mol oxirane) 45 g/L.
    floor slipperiness tall oil rosin; ethyoxylated fatty acid floor
ST
    slipperiness; alc ethyloxylated floor slipperiness prevention; castor oil
    ethoxylated floor slipperiness prevention; oleic acid ethyoxylated floor
    slipperiness prevention
ΙT
    Tall-oil rosin
    RL: USES (Uses)
        (emulsions, for floor slipperiness prevention)
TΤ
    Castor oil
    RL: USES (Uses)
        (ethyoxylated, emulsifiers, for tall-oil rosin emulsion
       manuf. for floor slipperiness prevention)
    Tall oil
IT
    RL: USES (Uses)
        (ethyoxylated, emulsifiers, for tall-oil rosin emulsion
       manuf., for floor slipperiness prevention)
IT
    Floors
        (slipperiness of, prevention of, tall-oil rosin emulsions
       for)
    Alcohols, compounds
TT
    RL: USES (Uses)
        (fatty, ethyoxylated, emulsifiers, for tall-oil rosin emulsion
       manuf., for floor slipperiness prevention)
ΙT
    Emulsifying agents
        (nonionic, ethyoxylated acids and alcs., for tall-oil resin
       emulsion, for floor slipperiness prevention)
               9004-96-0 25322-68-3D, ester and ether derivs
IT
    9002-92-0
    RL: USES (Uses)
        (emulsifiers, for tall-oil rosin emulsion manuf., for floor
       slipperiness prevention)
=> d 12 2-3 all
    ANSWER 2 OF 3 WPIDS COPYRIGHT 2004 THE THOMSON CORP on STN
   Full
  Text
    1993-313497 [40]
                       WPIDS
ΑN
```

DNC C1993-139255

Ink for jet printer providing images of high fastness to light and water -

consists of dispersed micronised pigment comprising continuous and

Emulsion increasing adhesion of shoe soles to floors

eb

DNN N1993-241413

TΙ

```
discontinuous liq. phases.
DC
     A97 E19 G02 T04
IN
     ARGENTERO, M; SOUDAZ, A M
     (OLIT) OLIVETTI & CO SPA
PA
CYC
    7
                     A1 19931006 (199340) * EN
                                                  6
                                                       C09D011-00
PI
     EP 564147
         R: DE FR GB NL
     JP 06025574
                                                  4
                                                       C09D011-00
                     A 19940201 (199409)
                                                  3
                     A 19941025 (199442)
                                                       C09D011-02
     US 5358555
                                                       C09D000-00
     IT 1259375
                     B 19960312 (199638)
     EP 564147
                                                  8
                     B1 19980513 (199823) EN
                                                       C09D011-00
         R: DE FR GB NL
     DE 69318465
                     E 19980618 (199830)
                                                       C09D011-00
     EP 564147 A1 EP 1993-302201 19930323; JP 06025574 A JP 1993-72071
ADT
     19930330; <u>US 5358555</u> A <u>US 1993-40252</u> 19930330; <u>IT 1259375</u> B <u>IT 1992-T0289</u>
     19920331; EP 564147 B1 EP 1993-302201 19930323; DE 69318465 E DE
     1993-618465 19930323, <u>EP 1993-3022</u>01 19930323
     DE 69318465 E Based on EP 564147
FDT
PRAI IT 1992-TO289
                          19920331
     DE 4211262; EP 249685; US 4692188; US 5047084
REP
IC
     C09D011-02
     ICM C09D000-00; C09D011-00; C09D011-02
AB
           564147 A UPAB: 19931129
     Pigmented ink for jet printers is a microdisperse 3- phase suspension
     comprising a discontinuous liq. phase (dtp), a continuous liq. phase (CLP)
     and a solid phase. Also new is a similar ink comprising hydrophobic and
     hydrophilic phases; a surfactant system; a biocide and an antioxidant.
          Pref. the solid phase is a pigment dispersed in DLP, which can be
     either hydrophilic or hydrophilic. The DLP is 2-30(3-10) wt.% of the ink
     which has viscosity 1-10 mPa.s.
          The surfactant system consists of an emulsifier (e.g. triethanolamine
     oleate or Na dodecyl sulphate); a co-emulsifier and a third surfactant as
     stabiliser (e.g. an ethoxylated alkylphenol, amine, alcohol or fatty
     acid; or an opt. ethoxylated ester of sorbitol).
          USE/ADVANTAGE - Ink is useful in bubble and thermal jet printer. It
     provides images of excellent fastness to water and light (i.e. they are
     indeliable) and show very low tendency to produce microprecipitates.
     Dwq.0/0
     CPI EPI
FS
FΑ
     AB; DCN
MC
     CPI: A12-W07D; E07-A02D; E07-D02; E10-A09A; E10-B02E; E10-B04D; E10-C04L;
          E10-E02E; E10-E03; E10-E04H; E10-E04M1; G02-A04A
     EPI: T04-G02C
L2
     ANSWER 3 OF 3 WPIDS COPYRIGHT 2004 THE THOMSON CORP on STN
   Full
   Text
     1980-82886C [47]
                        WPIDS
ΑN
     Cleaning compsns. contg. petroleum distillate and surfactant - for
ΤI
     cleaning hands, machinery, textiles, etc..
DC
PA
     (BARN-I) BARNES R I
CYC
                     A 19801113 (198047)*
     DE 3016707
PΙ
     GB 2049722
                     A 19801230 (198101)
     FR 2456137
                     A 19810109 (198109)
     JP 56004700
                     A 19810119 (198111)
                                                          19790619;
PRAI GB 1979-15743
                          19790505; GB 1979-21370
     GB 1980-12335
                          19800415
IC
     C11D003-16; C11D007-06
          3016707 A UPAB: 19930902
AB
     Cleaning compsns. comprise >=1 petroleum distillate and >=1 surfactant,
     pref. in the form of an emulsion with a gel-like consistency. Petroleum
     distillate can be a light mineral oil, a solvent, a hydrocarbon or a satd.
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or unsatd. vegetable or mineral oil (sic), pref. kerosene. The surfactant
    can be ionic or nonionic, pref. a polyalkoxylated alkylphenol, amine
     or alcohol, esp. polyoxyethylated octylphenol.
     Compsns. are non-acidic, non-alkaline and non-abrasive, and can be used to
     remove grease, soot, dirt, paint, etc., from hands, machinery, textiles
     and other surfaces.
FS
    CPI
    AB
FA
MC
    CPI: D11-A; D11-B16
=> d his
     (FILE 'HOME' ENTERED AT 15:31:36 ON 15 DEC 2004)
     FILE 'CAPLUS, WPIDS' ENTERED AT 15:31:48 ON 15 DEC 2004
L1
            110 S ALKYLPHENOL AMINE
L2
              3 S L1 AND (EMULSION OR MICROEMULSION OR EMULSIFIED)
=> s 11 and mannich
             6 L1 AND MANNICH
=> d 13 1-6 all
    ANSWER 1 OF 6 CAPLUS COPYRIGHT 2004 ACS on STN
          Citing
   Full
   Text
         References
     2003:678922 CAPLUS
ΑN
DN
     139:216749
ED
     Entered STN: 29 Aug 2003
    Amine-terminated polybutylene oxides and Mannich bases as gasoline
TΙ
     deposit inhibitors and detergents
    MacDuff, Malcolm G. J.; Chadwick, Richard C.
ΙN
PΑ
    The Lubrizol Corporation, USA
SO
     PCT Int. Appl., 23 pp.
    CODEN: PIXXD2
DT
    Patent
LA
    English
     ICM C10L001-22
IÇ
     ICS C10L010-00
     51-7 (Fossil Fuels, Derivatives, and Related Products)
CC
FAN.CNT 1
                                            APPLICATION NO.
     PATENT NO.
                         KIND
                                DATE
                                                                    DATE
                         ____
                          A2
                                            WO 2003-US4722
                                                                    20030218
     WO 2003070861
                                20030828
PΙ
                                20040219
     WO 2003070861
                         А3
            AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN,
             CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH,
             GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR,
             LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH,
             PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, TJ, TM, TN, TR, TT, TZ,
             UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
         RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
             IT, LU, MC, NL, PT, SE, SI, SK, TR
                                           EP 2003-709136
                                20041124
                                                                    20030218
    EP 1478717
                          Α2
             AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK
PRAI US 2002-357900P
                          Ρ
                                20020219
    WO 2003-US4722
                          W
                                20030218
CLASS
                CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
                ____
                        C10L001-22
WO 2003070861
                 TCM
                        C10L010-00
                 ICS
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h

eb c

g cg b

cg

```
AB
     Gasoline deposit inhibitors consist of: (1) a Mannich reaction product
     of a hydrocarbyl-substituted phenol with an aldehyde and an amine, and (2)
     a polyether amine of general structure RO(AO)mR1NR2R3, in which R =
     C8-30-hydrocarbyl, A = C2-6-alkylene, m = 1-50, R1 = C2-6-alkylene; and R2
     and R3 = H, hydrocarbyl, or -[R4N(R5)]nR6 (R4 = C2-6-alkylene; R5 and R6 =
     H or hydrocarbyl; and n = 1-7). Suitable Mannich reaction products are
     prepd. from an alkylphenol (esp. a polyisobutylene-phenol), formaldehyde,
     and ethylenediamine or dimethylamine. The additives can be added as a
     bulk treatment to the fuel (at 100-1000 wt. ppm) or an an aftermarket
     treatment to the fuel (at 1000-10,000 wt. ppm), in an engine operating
     under a clean-up cycle at engine speeds of ≥3000 rpm.
     gasoline deposit inhibitor detergent; Mannich base polyether amine
ST
     gasoline deposit inhibitor
IT
    Mannich bases
     RL: MOA (Modifier or additive use); SPN (Synthetic preparation); PREP
     (Preparation); USES (Uses)
        (additives; amine-terminated polybutylene oxides and Mannich
        bases as gasoline deposit inhibitors and detergents)
ΙT
     Polyoxyalkylenes, uses
     RL: MOA (Modifier or additive use); SPN (Synthetic preparation); PREP
     (Preparation); USES (Uses)
        (amine- and alkyl-terminated; amine-terminated polybutylene oxides and
        Mannich bases as gasoline deposit inhibitors and detergents)
ΙT
     Detergents
        (deposit inhibitors-detergents, as gasoline additives; amine-terminated
        polybutylene oxides and Mannich bases as gasoline deposit
        inhibitors and detergents)
     Gasoline additives
ΙT
        (deposit inhibitors-detergents; amine-terminated polybutylene oxides
        and Mannich bases as gasoline deposit inhibitors and
        detergents)
ΙT
     Gasoline additives
        (deposit inhibitors; amine-terminated polybutylene oxides and
        Mannich bases as gasoline deposit inhibitors and detergents)
ΙT
     Aldehydes, uses
     RL: MOA (Modifier or additive use); SPN (Synthetic preparation); PREP
     (Preparation); USES (Uses)
        (reaction products, with amines and alkylphenols;
        amine-terminated polybutylene oxides and Mannich
        bases as gasoline deposit inhibitors and detergents)
IT
     50-00-0DP, Formaldehyde, reaction products with amines and
                            95-48-7DP, o-Cresol, polyisobutenyl derivs.,
     polyisobutenylphenols
     reaction products with formaldehyde and ethylenediamine
     Ethylenediamine, reaction products with formaldehyde and
                            108-95-2DP, Phenol, polyisobutenyl derivs.,
     polyisobutenylphenols
     reaction products with formaldehyde and ethylenediamine (or dimethylamine)
     124-40-3DP, Dimethylamine, reaction products with formaldehyde and
     polyisobutenylphenols
     RL: MOA (Modifier or additive use); SPN (Synthetic preparation); PREP
     (Preparation); USES (Uses)
        (Mannich base detergents; amine-terminated polybutylene
        oxides and Mannich bases as gasoline deposit inhibitors and
        detergents)
ΙT
     325149-05-1P
                    587874-89-3P
                                   587874-90-6DP, \alpha-C8-30-hydrocarbyl
              587874-91-7DP, \alpha-C8-30-hydrocarbyl ethers
     RL: MOA (Modifier or additive use); SPN (Synthetic preparation); PREP
     (Preparation); USES (Uses)
        (detergents; amine-terminated polybutylene oxides and Mannich
        bases as gasoline deposit inhibitors and detergents)
IT
     291542-92-2P, Poly[oxy(ethyl-1,2-ethanediyl)], \alpha.-tridecyl-
     .ω.-hydroxy-
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
```

(synthesis and acrylonitrile reaction with; amine-terminated polybutylene oxides and **Mannich** bases as gasoline deposit inhibitors and detergents)

IT 587874-92-8P 587875-16-9P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(synthesis and hydrogenation of; amine-terminated polybutylene oxides and Mannich bases as gasoline deposit inhibitors and detergents)

L3 ANSWER 2 OF 6 CAPLUS COPYRIGHT 2004 ACS on STN

Full Citing References Text 1985:598427 CAPLUS AN DN 103:198427 ED Entered STN: 14 Dec 1985 Middle distillate containing storage stability additive TISung, Rodney L.; Karol, Thomas J. IN Texaco Inc. , USA PA U.S., 8 pp. SO CODEN: USXXAM DT Patent LA English IC ICM C10L001-22 NCL 044073000 51-9 (Fossil Fuels, Derivatives, and Related Products) CC FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE ______ ____ ____. _____ US 4533361 Α 19850806 US 1984-658933 19841009 PΙ PRAI US 1984-658933 19841009 CLASS CLASS PATENT FAMILY CLASSIFICATION CODES PATENT NO. ICM C10L001-22 US 4533361 NCL 044073000

R - NR 1(NR 1A) NACR 2R 30H

GΙ

The storage stability of middle distillates, such as diesel fuel, was improved by the addn. of acylated condensates of an alkenylsuccinic anhydride and a Mannich base [I, where R is an unsatd. radical (from a polyolefin), R1 is a divalent org. radical, R2 is H, alkyl, aryl, cycloalkyl, or combination, R3 is arylene, A is the same as R1 or an acyl group, and a = 1-20]. Thus, heating of 2,6-di-tert-butylphenol 0.167, HCHO 0.25, polyisobutenylsuccinic anhydride 0.185, oxalic acid 0.527, and pentaethylenehexamine 0.167 mol in the presence of diluent oil and a silicone antifoaming agent at 110-160° for ~9 h gave an additive which, at 25 lb/1000 bbl diesel oil concn., decreased its sedimentation rating (Potential Deposit Test) to 1.

ST antisedimentation additive diesel fuel; stabilizer storage middle petroleum distillate

IT Fuels, diesel

(storage of, antisedimentation additives for)

I.

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IT Petroleum products
```

(middle distillates, storage of, antisedimentation additives for)

50-00-0D, condensation products with alkenylsuccinic anhydride, IT 57-13-6D, condensation products with alkylphenols, and amines, acylated alkenylsuccinic anhydride, alkylphenols, formaldehyde, and polyamines 79-14-1D, condensation products with alkenylsuccinic anhydride, 108-30-5D, alkylphenols, amines, and formaldehyde polyisobutenyl derivs., condensation products with alkylphenols and amines 112-57-2D, condensation products with and formaldehyde, acylated alkenylsuccinic anhydride, alkylphenols, and formaldehyde, acylated 144-62-7D, condensation products with alkenylsuccinic anhydride, alkylphenols, amines, and formaldehyde 4067-16-7D, condensation products with alkenylsuccinic anhydride, alkylphenols, and 26746-38-3D, condensation products with formaldehyde, acylated alkenylsuccinic anhydride, amines, and formaldehyde, acylated RL: USES (Uses) (antisedimentation additives, for storage of petroleum middle

distillates)

ANSWER 3 OF 6 CAPLUS COPYRIGHT 2004 ACS on STN

Full Citing Text References

AN 1983:490697 CAPLUS

DN 99:90697

L3

ED Entered STN: 12 May 1984

- TI Production of ashless additives from technical alkylsalicylic acids
- AU Korotushenko, T. P.; Sukhoverkhov, V. D.; Gordash, Yu. T.; Grechko, A. N.
- CS VNIIPKneftekhim., Kiev, USSR
- SO Neftepererabotka i Neftekhimiya (Kiev) (1983), 24, 22-5 CODEN: NEFNBY; ISSN: 0548-1406
- DT Journal
- LA Russian
- CC 51-8 (Fossil Fuels, Derivatives, and Related Products)
- AB Condensation of diethylenetriamine or polyethylenepolyamine with a 54.2:45.8 mixt. of tech. alkylsalicylic acids and alkylphenols at 180° (optimum temp.) gave the corresponding amide-alkylphenol mixt. This mixt. was treated with an aq. HCHO soln. at 90-110° to give Mannich bases for use as lubricating oil additives. The diethylenetriamine-alkylsalicylic acid-alkylphenol-HCHO condensation product had good detergent and corrosion-inhibiting properties.
- ST lubricating oil detergent Mannich base; corrosion inhibitor Mannich base lubricant
- IT Mannich bases

RL: USES (Uses)

(corrosion inhibitors-detergents, for lubricating oils)

IT Lubricating oil additives

(corrosion inhibitors, Mannich bases as)

IT Amines, compounds

RL: USES (Uses)

(polyethylenepoly-, reaction products with alkylphenols and alkylsalicylic acids and formaldehyde, corrosion inhibitors-detergents, for lubricating oils)

IT 69-72-7D, alkyl derivs., reaction products with alkylphenols, amines, and formaldehyde 108-95-2D, alkyl derivs., reaction products with alkylsalicylamides and formaldehyde 111-40-0D, reaction products with alkylphenols and alkylsalicylic acids and formaldehydes RL: USES (Uses)

(corrosion inhibitors-detergents, for lubricating oils)

L3 ANSWER 4 OF 6 CAPLUS COPYRIGHT 2004 ACS on STN

Full Citing Text References

AN 1978:107582 CAPLUS

DN 88:107582

- ED Entered STN: 12 May 1984
- TI Reaction of phosphorus pentasulfide with hydroxyl-containing compounds in relation to the synthesis of additive INKhP-21
- AU Seleznenko, L. V.
- CS Vses. Nauchno-Issled. Proektno-Konstr. Inst. Neftepererab. Neftekhim. Prom., Kiev, USSR
- SO Neftepererabotka i Neftekhimiya (Kiev) (1977), 15, 35-9 CODEN: NEFNBY; ISSN: 0548-1406
- DT Journal; General Review
- LA Russian
- CC 51-0 (Fossil Fuels, Derivatives, and Related Products) Section cross-reference(s): 22
- AB A review, with 24 refs., of the mechanism and kinetics of reaction of P2S5 with a Mannich base obtained by condensation of alkylphenols, amines, and HCHO. This reaction is basic in the prepn. of the antioxidant INKhP 21 [12798-12-8] for lubricating oils.
- ST review mechanism phosphorus pentasulfide; **Mannich** base prepn kinetics review; antioxidant lubricating oil prepn review
- IT Kinetics, reaction

(of phosphosulfurization of **Mannich** bases, in prepn. of lubricating oil additives)

IT Mannich bases

RL: RCT (Reactant); RACT (Reactant or reagent) (reaction of, with phosphorus sulfide in prepn. of lubricating oil additives, kinetics and mechanism of)

IT Lubricating oil additives

(antioxidants, barium salts of phosphosulfurized phenolic resins, prepn. of, phosphosulfurization in)

IT Phenolic resins, compounds

(phosphosulfurized, barium salts, lubricating oil antioxidants, prepn. of, phosphosulfurization in)

IT <u>108-95-2DP</u>, alkyl derivs., polymer with ammonia and formaldehyde, phosphosulfirized, barium salts <u>12798-12-8P</u>

RL: PREP (Preparation)

(lubricating oil antioxidants, prepn. of, phosphosulfurization in)

IT 1314-80-3

RL: RCT (Reactant); RACT (Reactant or reagent)
(reaction of, with **Mannich** bases in prepn. of lubricating oil additives, kinetics and mechanism of)

L3 ANSWER 5 OF 6 CAPLUS COPYRIGHT 2004 ACS on STN

Full Citing Text References

- AN 1976:601747 CAPLUS
- DN 85:201747
- ED Entered STN: 12 May 1984
- TI Use of thin-layer chromatography for monitoring the condensation stages of alkylphenols with formaldehyde and amines
- AU Sopkina, A. K.; Marusyak, O. V.; Gordash, Yu. T.; Zhurba, A. S.
- CS Vses. Nauchno-Issled. Proektn. Inst. Kompleksn. Neftekhim. Protsessov, Leningrad, USSR
- SO Khimiya i Tekhnologiya Topliv i Masel (1976), (5), 53-5 CODEN: KTPMAG; ISSN: 0023-1169
- DT Journal
- LA Russian
- CC 80-4 (Organic Analytical Chemistry)
- AB The products of condensation of alkylphenols with formaldehyde and amines were analyzed by thin-layer chromatog. on Al2O3 by using C6H6-MeOH (9:1) as developing solvent, iodine vapor as detection reagent, and 5-20 μg sample in C6H6 soln. The anal. took 30-40 min. The end product was a Mannich base (I). The chromatogram obtained during condensation of C11-13 2,6-dialkylphenol (II) with HCHO and an amine showed that a small amt. of I was formed by conversion of monoalkylphenols present in the original dialkylphenol and not from II. Similarly, the chromatogram

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obtained when controlling the removal of unreacted diethylenetriamine
     (III) from the condensation products indicated the end of the washing
    procedure by the disappearance of the spot of III. The method was simple
     and rather reliable for qual. control, giving also semiquant. data.
     thin layer chromatog alkylphenol condensation; formaldehyde condensation
ST
     alkylphenol analysis; amine condensation alkylphenol analysis
ΙT
    Amines, reactions
    RL: RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent)
        (condensation of, with alkylphenols, thin-layer chromatog. monitoring
ΙT
     Chromatography, thin-layer
        (of condensation products of alkylphenols amines
        and formaldehyde)
IT
     50-00-0, reactions
     RL: RCT (Reactant); ANST (Analytical study); RACT (Reactant or reagent)
        (condensation of, with alkylphenols, thin-layer chromatog. monitoring
        of)
ΙT
     108-95-2D, Phenol, alkyl derivs.
     RL: ANST (Analytical study)
        (condensation of, with amines and formaldehyde, thin-layer chromatog.
        monitoring of)
     ANSWER 6 OF 6 WPIDS COPYRIGHT 2004 THE THOMSON CORP on STN
L3
   Full
     1979-21913B [11]
                      WPIDS
AN
    Mannich reaction prods. useful as lubricating oil detergent additives -
TΙ
     prepd. from carboxylic acid-substd. alkylphenol, an amine and
     formaldehyde.
DC
     A97 H07
     KARLL, R E; LEE, R J
ΙN
     (STAD) STANDARD OIL CO INDIANA
PΑ
CYC 1
     US 4142980
                     A 19790306 (197911)*
ΡI
PRAI US 1977-831732
                          19770909; <u>US 1978-945316</u>
                                                          19780925
     C07C101-72; C10M001-32; C10M003-26; C10M005-20
IC
          4142980 A UPAB: 19930901
AΒ
     An oil soluble cpd. (I) is claimed which is the Mannich reaction prod.
     of (a) the reaction prod. (II) of a (50-20000C alkyl)) phenol and a 2-100C
     aliphatic unsatd. carboxylic acid, (b) a <=100C amine, having >=1 reactive
     N of formula H2N-(AlNH)xH (IIIe (where A is 2-6C alkylene and x is 1-10),
     and (c) formaldehyde or a formaldehyde-affording reactant (IV), the ratio
     (II):(III):(IV) being 1:0.7-1.0:1.5-2.
          (I) are very effective in small amts. (0.05-5 wt.%, pref. 0.1-2 wt.%)
     as detergent/dispersant additives for lubricating oils.
FS
    CPI
FΑ
    AΒ
     CPI: A10-E03; A10-E19; A12-W02A; H07-G03
MC
=> d his
     (FILE 'HOME' ENTERED AT 15:31:36 ON 15 DEC 2004)
     FILE 'CAPLUS, WPIDS' ENTERED AT 15:31:48 ON 15 DEC 2004
L1
            110 S ALKYLPHENOL AMINE
L2
              3 S L1 AND (EMULSION OR MICROEMULSION OR EMULSIFIED)
L3
              6 S L1 AND MANNICH
=> s mannich and (emulsion or microemulsion or emulsified)
           149 MANNICH AND (EMULSION OR MICROEMULSION OR EMULSIFIED)
=> s 14 and (fuel or gasoline or diesel)
            10 L4 AND (FUEL OR GASOLINE OR DIESEL)
L5
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g cg b

cg

eb c

h

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L5 ANSWER 1 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN
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Citing
         References
   Text
    2004:3298
               CAPLUS
AN
DN
     140:62092
ED
    Entered STN: 04 Jan 2004
    Method of improving the compatibility of a fuel additive composition
ΤI
    containing a Mannich condensation product
    Carabell, Kevin D.; Gray, James A.
IN
    Chevron Oronite Co., LLC, USA
PA
    U.S. Pat. Appl. Publ., 12 pp.
SO
    CODEN: USXXCO
.
DT
    Patent
LA
    English
IC
     ICM C10L001-18
     ICS C10L001-24; C10L001-22
NCL
     044330000
     51-9 (Fossil Fuels, Derivatives, and Related Products)
CC
     Section cross-reference(s): 21, 27
                                                                   DATE
                                            APPLICATION NO.
                       KIND DATE
     PATENT NO.
                                                                   20020618
                                            US 2002-175143
                        A1
                                20040101
PΙ
     US 2004000089
     US 6733551
                         В2
                                20040511
                        A2
                                            EP 2003-252535
                                                                   20030423
                                20040102
     EP 1375629
                         A3
                                20040114
     EP 1375629
             AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK
                                            CA 2003-2428909
                                                                    20030516
     CA 2428909
                          AΑ
                                20031218
                                20020618
PRAI US 2002-175143
                          Α
CLASS
                 CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
                ____
                        C10L001-18
                 ICM
 US 2004000089
                        C10L001-24; C10L001-22
                 ICS
                 NCL
                        044330000
                        C10L001/14B; C10L001/22W; C10L010/00; C10M159/16;
 US 2004000089
                 ECLA
                        C10M167/00
                        C10L001/14B; C10L001/22W; C10L010/00; C10M159/16;
 EP 1375629
                 ECLA
                        C10M167/00
     MARPAT 140:62092
OS
     A method of improving the compatibility of a fuel additive compn. is
AΒ
     comprised of blending together the following components: (a) a Mannich
     condensation product of (1) a high mol. wt. alkyl-substituted hydroxyarom.
     compd., (2) an amine having the formula: 1 wherein A is CH or nitrogen, R1
     , R2 , R3 are independently hydrogen or lower alkyl of C1-C~6 and each R2
     and R3 is independently selected in each -CR2R3 - unit, and x is an
     integer from 1 to ~6; and (3) an aldehyde, wherein the resp. molar
     ratio of reactants (1), (2), and (3) is 1:0.1-2:0.1-2; (b) a
     hydrocarbyl-terminated poly(oxyalkylene) mono-ol; (c) a carboxylic acid as
     represented by the formula: R4(COOH)y, wherein R4 represents a C2 -~50
     hydrocarbyl group, and y represents an integer of 1 to ~4; and (d)
     an anhydride selected from the group consisting of succinic, glutaric,
     phthalic, and alkyl anhydrides.
     fuel additive Mannich condensation carboxylic acid anhydride
ST
     polyoxyalkylene ether
     Carboxylic acids, uses
IT
     RL: MOA (Modifier or additive use); USES (Uses)
        (C3-C51; method of improving the compatibility of fuel
        additive compn. contg. Mannich condensation product and
        carboxylic acids)
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ΙT
    Carboxylic acids, uses
    RL: MOA (Modifier or additive use); USES (Uses)
        (C6-C54 tetracarboxylic acids; method of improving the compatibility of
        fuel additive compn. contg. Mannich condensation
        product and carboxylic acids)
    Alcohols, uses
ΙT
    RL: MOA (Modifier or additive use); USES (Uses)
        (alkoxylated; method of improving the compatibility of fuel
        additive compn. contg. Mannich condensation product and
        carboxylic acids)
     Polyoxyalkylenes, uses
TΤ
     RL: MOA (Modifier or additive use); USES (Uses)
        (alkyl group-terminated, propoxylated and butoxylated; method of
        improving the compatibility of fuel additive compn. contg.
        Mannich condensation product and carboxylic acids)
     Phenols, uses
ΙT
    RL: MOA (Modifier or additive use); USES (Uses)
        (alkyl, C21-C356 alkyl, Mannich condensation reaction
        products with aldehydes and amines; method of improving the
        compatibility of fuel additive compn. contg. Mannich
        condensation product and carboxylic acids)
    Aromatic hydrocarbons, uses
ΙT
     RL: TEM (Technical or engineered material use); USES (Uses)
        (alkyl, light alkylate; method of improving the compatibility of
        fuel additive compn. contg. Mannich condensation
        product and carboxylic acids)
IT
     Corrosion inhibitors
        (carboxylic acid type; method of improving the compatibility of
        fuel additive compn. contg. Mannich condensation
        product and carboxylic acids)
ΙT
     Fuel oil additives
        (compatibilizers; method of improving the compatibility of fuel
        additive compn. contg. Mannich condensation product and
        carboxylic acids)
IT
    Emulsions
        (demulsifiers; method of improving the compatibility of fuel
        additive compn. contg. Mannich condensation product and
        carboxylic acids)
     Phenols, reactions
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (derivs.; method of improving the compatibility of fuel
        additive compn. contg. Mannich condensation product and
        carboxylic acids)
ΙT
     Amines, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (diamines; method of improving the compatibility of fuel
        additive compn. contg. Mannich condensation product and
        carboxylic acids)
ΙT
     Carboxylic acids, uses
     RL: MOA (Modifier or additive use); USES (Uses)
        (dicarboxylic, C4-C52; method of improving the compatibility of
        fuel additive compn. contq. Mannich condensation
        product and carboxylic acids)
     Polyoxyalkylenes, uses
IT
     RL: MOA (Modifier or additive use); USES (Uses)
        (ethers, hydrocarbyl-terminated mono-ols; method of improving the
        compatibility of fuel additive compn. contg. Mannich
        condensation product and carboxylic acids)
ΙT
     Polyoxyalkylenes, uses
     RL: MOA (Modifier or additive use); USES (Uses)
        (ethers, with C7-C30 alkyl phenols; method of improving the
        compatibility of fuel additive compn. contg. Mannich
        condensation product and carboxylic acids)
     Fuel oil additives
IT
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(method of improving the compatibility of fuel additive
   compn. contg. Mannich condensation product and carboxylic
   acids)
Anhydrides
RL: MOA (Modifier or additive use); USES (Uses)
   (method of improving the compatibility of fuel additive
   compn. contg. Mannich condensation product and carboxylic
   acids)
Mannich bases
RL: MOA (Modifier or additive use); USES (Uses)
   (phenolic, condensation products of alkylphenols with aldehydes with
   piperazine-based triamines or piperidine-based diamines; method of
   improving the compatibility of fuel additive compn. contg.
   Mannich condensation product and carboxylic acids)
Mannich bases
RL: MOA (Modifier or additive use); PRP (Properties); PUR (Purification or
recovery); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)
   (phenolic, condensation products of polyisobutenylphenol with
   paraformaldehyde and 1-(2-aminoethyl)piperazine; method of improving
   the compatibility of fuel additive compn. contg.
   Mannich condensation product and carboxylic acids)
Phenols, uses
RL: MOA (Modifier or additive use); USES (Uses)
   (polypropyl and polyisobutyl derivs., Mannich condensation
   reaction products with aldehydes and amines; method of improving the
   compatibility of fuel additive compn. contg. Mannich
   condensation product and carboxylic acids)
Sedimentation (separation)
   (redn. of; method of improving the compatibility of fuel
   additive compn. contg. Mannich condensation product and
   carboxylic acids)
Stability
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IT

ΙT

IT

IT

IT

TΤ

(to air oxidn.; method of improving the compatibility of fuel additive compn. contg. Mannich condensation product and carboxylic acids)

IT Amines, reactions

> RL: RCT (Reactant); RACT (Reactant or reagent) (triamines; method of improving the compatibility of fuel additive compn. contg. Mannich condensation product and carboxylic acids)

TΤ Carboxylic acids, uses

> RL: MOA (Modifier or additive use); USES (Uses) (tricarboxylic acids, C5-C53; method of improving the compatibility of fuel additive compn. contg. Mannich condensation product and carboxylic acids)

7732-18-5, Water, processes IT

RL: FMU (Formation, unclassified); REM (Removal or disposal); FORM (Formation, nonpreparative); PROC (Process) (method of improving the compatibility of fuel additive compn. contq. Mannich condensation product and carboxylic

ΙT 50-00-0D, Formaldehyde, condensation reaction products with alkylphenols and piperidine- and piperazine- based alkylamine and dialkylamine derivs. 108-30-5, Succinic anhydride, uses 85-44-9, Phthalic anhydride 108-30-5D, Succinic anhydride, derivs. 108-55-4, Glutaric anhydride 110-85-0D, Piperazine, alkylaminoalkyl derivs., Mannich condensation reaction products with alkylphenols and aldehydes 110-89-4D, Piperidine, alkylaminoalkyl derivs., Mannich condensation reaction products with alkylphenols and aldehydes 140-31-8D, 1-Piperazineethanamine, condensation Oleic acid, uses 140-31-8D, reaction products with alkylphenols and aldehydes 1-Piperazineethanamine, α and/or β - mono- and poly- Me and Et derivs. condensation reaction products with alkylphenols and aldehydes, 26544-38-7, Tetrapropenylsuccinic anhydride 30525 - 89 - 4D

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Paraformaldehyde, condensation reaction products with alkylphenols and
     piperidine- and piperazine- based alkylamine and dialkylamine derivs.
     76025-62-2D, 4-Piperidineethanamine, condensation reaction products with
                                  76025-62-2D, 4-Piperidineethanamine, \alpha
     alkylphenols and aldehydes
     and/or \beta- mono- and poly- Me and Et derivs., condensation reaction
     products with alkylphenols and aldehydes
     RL: MOA (Modifier or additive use); USES (Uses)
        (method of improving the compatibility of fuel additive
        compn. contg. Mannich condensation product and carboxylic
        acids)
     108-95-2D, Phenol, Polyisobutenyl derivs.
                                                  140-31-8, 1-(2-
ΙT
     Aminoethyl)piperazine
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (method of improving the compatibility of fuel additive
        compn. contq. Mannich condensation product and carboxylic
        acids)
IT
     29011-16-3, Polypropylene glycol monopropyl ether
     RL: MOA (Modifier or additive use); USES (Uses)
        (synthetic carrier fluid; method of improving the compatibility of
        fuel additive compn. contg. Mannich condensation
        product and carboxylic acids)
L5
     ANSWER 2 OF 10
                    CAPLUS COPYRIGHT 2004 ACS on STN
          Citing
   Full
         References
AN
     2002:143100 CAPLUS
     136:202883
DN
     Entered STN: 22 Feb 2002
ED
     Amino alkylphenol emulsifiers for an aqueous hydrocarbon fuel
ΤI
     Filippini, Brian B.; Forsberg, John W.; Steckel, Thomas F.; Moreton, David
IN
     J.; Mcatee, Rodney J.
PΑ
     USA
     U.S. Pat. Appl. Publ., 14 pp., Cont.-in-part of U.S. Ser. No. 483,481.
SO
     CODEN: USXXCO
DT
     Patent
LA
     English
IC
     ICM C10L001-32
NCL
    044301000
     51-7 (Fossil Fuels, Derivatives, and Related Products)
CC
FAN.CNT 13
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|------------|----------------|----------------------------------|-----|-----|-----|-------------|----------|----------|------|-----------------|-----------------|-----|-----|------|----------|----------|-----|-----|--|
| | PATENT NO. | | | | | | DATE | | | APPLICATION NO. | | | | DATÉ | | | | | |
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| PΙ | US 2002020106 | | | | | A1 | | 20020221 | | | US 2001-892073 | | | | 20010626 | | | | |
| | US | US 6368366 | | | | B1 | | 20020409 | | | US 1999-349268 | | | | | 19990707 | | | |
| | US | JS 6368367 | | | | В1 | | 20020409 | | | US 1999-390925 | | | | | 19990907 | | | |
| | US | IS 6383237 | | | | В1 | | 20020507 | | | US 2000-483481 | | | | | 20000114 | | | |
| | US | US 2001020344 | | | | A1 | | 20010913 | | | US 2001-755577 | | | | 20010105 | | | | |
| | WO | NO 2003002693 | | | | A2 | | 2003 | 0109 | | WO 2002-US18405 | | | | | 20020611 | | | |
| | WO | | | | | A3 20030821 | | | | | | | | | | | | | |
| | | W: | AE, | AG, | AL, | AM, | AT, | AU, | AZ, | BA, | BB, | BG, | BR, | BY, | ΒZ, | CA, | CH, | CN, | |
| | | | co, | CR, | CU, | CZ, | DE, | DK, | DM, | DZ, | EC, | EE, | ES, | FI, | GB, | GD, | GE, | GH, | |
| | | | | | | | | | | | KE, | | | | | | | | |
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| | | RW: | | | - , | • | , | • | | | FR, | GB, | GR, | ΙE, | IT, | LU, | MC, | NL, | |
| | | | | SE, | | · | · | · | • | | | | • | | | | | | |
| PRAI | US | US 1999-349268 | | | | A2 19990707 | | | | | | | | | | | | | |
| | | US 1999-390925
US 2000-483481 | | | | | | 1999 | 0907 | | | | | | | | | | |
| | | | | | | | | 20000114 | | | | | | | | | | | |
| | _ | US 2001-755577 | | | | | | | 0105 | | | | | • | | | | | |
| | US 2001-892073 | | | | A | | 20010626 | | | | | | | | | | | | |
| CLASS | | | | | | | | | | | | | | | | | | | |
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PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES

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US 2002020106
                 ICM
                        C10L001-32
                 NCL
                        044301000
 US 2002020106
                 ECLA
                        C10L001/32D
 US 6368366
                 ECLA
                        C10L001/32D
 US 6368367
                 ECLA
                        C10L001/32D
 US 6383237
                 ECLA
                        C10L001/32D
 US 2001020344
                 ECLA · C10L001/32D
     MARPAT 136:202883
OS
     Novel animoalkylphenol emulsifiers are used for making aq. hydrocarbon
AB
     fuel emulsions suitable for engines. The emulsifier comprises (a) an
     aminoalkylphenol which is made by reacting alkylphenol (e.g.,
     polypropylphenol, polybutylphenol, poly(isopropanol)phenol,
     polyamylphenol, tetrapropylphenol, or substituted phenols), an aldehyde
     (e.g., formaldehyde, acetaldehyde, or aldol) and an amine (e.g.,
     alkanolamines, polyalkylene amines, etc.) resulting in an amino alkyl
     phenol.
ST
     fuel water emulsifier
ΙT
     Emulsifying agents
       Mannich reaction
        (amino alkylphenol emulsifiers for an aq. hydrocarbon fuel)
ΙT
        (emulsifiers; amino alkylphenol emulsifiers for an ag. hydrocarbon
        fuel)
ΙT
     Fuels
        (oil-in-water; amino alkylphenol emulsifiers for an aq. hydrocarbon
                                  106-50-3, p-Phenylene diamine, reactions
ΙT
     78-90-0, Propylene diamine
     108-95-2D, Phenol, isobutylene derivs., polymers
                                                         109-55-7
     Pyrrole 110-89-4, Piperidine, reactions 110-91-8, Morpholine,
                 111-40-0, Diethylenetriamine
                                                111-42-2, Diethanolamine,
     reactions
                 112-24-3, Triethylene tetramine
                                                    112-57-2, Tetraethylene
     reactions
                                              123-00-2, 4-
                 120-72-9, Indole, reactions
     pentamine
                                                                 123-90-0,
                            <u>123-75-1</u>, Pyrrolidine, reactions
     Morpholinepropanamine
                                               <u>288-13-1</u>, Pyrazole <u>288-32-4</u>,
                      135-67-1, Phenoxazine
     Thiomorpholine
                            289-95-2D, Pyrimidine, N-alkyl amine derivs.
     Imidazole, reactions
                              504-74-5, Imidazolidine
                                                         504 - 74 - 5D
     504-70-1, Pyrazolidine
     Imidazolidine, N-alkyl amine derivs. 6484-52-2, Ammonium nitrate,
                 28299-33-4, Imidazoline 28350-87-0, Pyrroline
     reactions
                                                                    36118-45-3,
                  36354-95-7, N-Acetyl tetraethylenepentamine
                                                                 38096-30-9,
     Pyrazoline
     Diamino naphthalene
                           58437-91-5D, reaction products with alkylamines
     301356-20-7
                  400771-77-9
                                 400771-78-0
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (amino alkylphenol emulsifiers for an aq. hydrocarbon fuel)
L5
     ANSWER 3 OF 10 WPIDS COPYRIGHT 2004 THE THOMSON CORP on STN
   Full
   Text
ΑN
     2004-591760 [57]
                        WPIDS
     2004-088866 [09]
CR
                        DNC C2004-215146
DNN
    N2004-467980
     Reduction of engine wear during operation of internal combustion engine
ΤI
     involves recirculating exhaust gas from engine to intake air supply of
     engine; and operating engine using water-blended fuel composition.
DC
     A18 A88 A95 Q52 Q53
IN
     DUNCAN, DA; LANGER, DA; SHAH, MP; ZALAR, FV
     (DUNC-I) DUNCAN D A; (LANG-I) LANGER D A; (SHAH-I) SHAH M P; (ZALA-I)
PA
     ZALAR F V; (LUBR) LUBRIZOL CORP
CYC
     US 2004139931
PΙ
                     A1 20040722 (200457)*
                                                16
                                                       F02M025-07
     US 6823822
                     B2 20041130 (200479)
                                                      F02B047-00
    <u>US 2004139931</u> A1 Div ex <u>US 2002-90500</u> 20020304, US 2004-754269 20040109;
ADT
     US 6823822 B2 Div ex US 2002-90500 20020304, US 2004-754269 20040109
    US 6823822 B2 Div ex US 6748905
FDT
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PRAI US 2002-90500 20020304; US 2004-754269 20040109

C ICM F02B047-00; F02M025-07

ICS F02B047-08

AB US2004139931 A UPAB: 20041208

NOVELTY - Reduction of engine wear in the operation of an internal combustion engine involves: recirculating at least part of the exhaust gas from the engine to the intake air supply of the engine; and operating the engine using a water-blended **fuel** composition, which is obtained by combining a normally liquid hydrocarbon **fuel**, water and at least one surfactant

DETAILED DESCRIPTION - Reduction of engine wear in the operation of an internal combustion engine involves:

- (A) recirculating at least part of the exhaust gas from the engine to the intake air supply of the engine; and
 - (B) operating the engine using a water-blended fuel composition (C1).
- (C1) Is obtained by combining a normally liquid hydrocarbon fuel (preferably diesel fuel), water and at least one surfactant. The surfactant comprises at least one product (P1), at least one product (P2), at least one mannich reaction product (P3) and/or at least one ionic or a nonionic compound (P4) having a hydrophilic-lipophilic balance of 1 40. (P1) Is made from the reaction of an acylating agent and ammonia, amine, hydroxyamine and/or alcohol. (P2) Is derived from a polycarboxylic acylating agent (A1), a copolymer (A2), and linking compound (A3). (A2) Is derived from at least one of olefin monomer (preferably alpha-olefin) and alpha, beta unsaturated carboxylic acid (preferably maleic anhydride) or its derivative. (A3) Has at least two (preferably at least one) of primary amino, secondary amino and hydroxy group (preferably ethylene polyamine). (P3) Is derived from hydroxy aromatic compound, aldehyde or ketone, and amine containing at least one primary or secondary amino group.

USE - For reducing engine wear comprising piston ring wear reduction or cylinder linear wear reduction in the operation of compression ignition engine equipped with exhaust gas recirculation system (claimed).

 ${\tt ADVANTAGE}$ - The process reduces the generation of nitric oxide and particulate emission in the exhaust of the engine.

Dwg.0/4

FS CPI GMPI

FA AB

MC CPI: A10-E01; A12-T03B

L5 ANSWER 4 OF 10 WPIDS COPYRIGHT 2004 THE THOMSON CORP on STN

Full Text

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AN 2004-224195 [21] WPIDS
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CR 2004-063955 [07]; 2004-224194 [21]

DNC C2004-088421

TI Friction modifier for use in **fuel**, e.g. **gasoline** for internal combustion engine, comprises alkoxylated amine salt or etheramine salt of saturated carboxylic acid.

DC A95 D22 E19 H06

IN ARADI, A A; COLUCCI, W J; MALFER, D J; SCHWAB, S D

PA (ETHY) ETHYL CORP; (ARAD-I) ARADI A A; (COLU-I) COLUCCI W J; (MALF-I) MALFER D J; (SCHW-I) SCHWAB S D

CYC 34

PI US 2004010967 A1 20040122 (200421)* 11 C10L001-22 EP 1471131 A1 20041027 (200471) EN C10L001-22

R: AL AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HU IE IT LI LT LU LV

MC MK NL PT RO SE SI SK TR

<u>CA 2436194</u> A1 20041022 (200474) EN C10L001-22 JP 2004323821 A 20041118 (200476) 27 C10M129-32

ADT US 2004010967 A1 CIP of US 2002-128529 20020424, US 2003-421006 20030422; EP 1471131 A1 EP 2003-17294 20030730; CA 2436194 A1 CA 2003-2436194

20030729; <u>JP 2004323821</u> A <u>JP 2003-292448</u> 20030812 <u>PRAI US 2003-421006</u> 20030422; <u>US 2002-128529</u> 20020424

IC ICM C10L001-22; C10M129-32

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ICS C10L001-14; C10L001-18; C10M129-34; C10M129-40; C10M129-42;
          C10M129-44; C10M133-06; C10M133-08; C10M141-06
AB
     US2004010967 A UPAB: 20041125
     NOVELTY - A friction modifier comprises alkoxylated amine salt or
     etheramine salt of saturated carboxylic acid.
          DETAILED DESCRIPTION - INDEPENDENT CLAIMS are also included for:
          (a) an additive concentrate composition comprising saturated
     carboxylic acid salt of alkoxylated amine, and detergent package
     containing detergent and carrier fluid;
          (b) a fuel composition comprising combustible fuel and additive
     concentrate (50-2500 ppm);
          (c) a method of combusting a fuel by combining a fuel which is
     combustible in a gasoline engine having intake valves with a fuel
     additive concentrate; and
          (d) a method of increasing the fuel efficiency in a gasoline
     combustion engine by combusting a gasoline fuel comprising fuel
     boiling in the gasoline boiling range, and saturated carboxylic acid
     salt of alkoxylated amine or etheramine.
          USE - For use in fuel, e.g. gasoline for internal combustion
     engine (claimed).
          ADVANTAGE - The inventive friction modifier enables the formulation
     of stable additive concentrate that provides benefit in friction loss when
     incorporated in fuel. The additive concentrate increases fuel
     efficiency without increasing the incidence of intake valve deposits in
     combustion engine.
     Dwq.0/0
FS
     CPI
FΑ
     AB; DCN
     CPI: A12-T03A; D09-B; E07-D03; E10-B03B; E10-C04L; E10-G02H2; H06-B01;
MC
          H06-D06
     ANSWER 5 OF 10 WPIDS COPYRIGHT 2004 THE THOMSON CORP on STN
L5
   Full
   Text
     2004-088866 [09]
                        WPIDS
AN
CR
     2004-591760 [57]
                        DNC C2004-036193
DNN N2004-071130
     Process to reduce engine wear in internal combustion engine comprises
TI
     recirculating part of exhaust gas from the engine to intake air supply and
     operating engine using water-blended fuel composition.
DC
     A14 A17 A25 A95 E19 H06 Q52
     DUNCAN, D A; LANGER, D A; SHAH, M P; ZALAR, F V
ΙN
     (LUBR) LUBRIZOL CORP; (DUNC-I) DUNCAN D A; (LANG-I) LANGER D A; (SHAH-I)
     SHAH M P; (ZALA-I) ZALAR F V
CYC
     US 2003164147
                     A1 20030904 (200409)*
                                                16
                                                       F02B047-00
PΙ
     WO 2003076783
                     A2 20030918 (200409)
                                           ΕN
                                                       F02D021-00
        RW: AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HU IE IT LU MC NL PT SE
         W: AE AG AL AM AT AU AZ BA BB BG BR BY BZ CA CH CN CO CR CU CZ DE DK
            DM DZ EC EE ES FI GB GD GE GH GM HR HU ID IL IN IS JP KE KG KP KR
            KZ LC LK LR LS LT LU LV MA MD MG MK MN MW MX MZ NO NZ OM PH PL PT
            RO RU SC SD SE SG SK SL TJ TM TN TR TT TZ UA UG UZ VC VN YU ZA ZM
            ZW
                     A1 20030922 (200431)
                                                      F02D021-00
     AU 2003225593
                     B2 20040615 (200439)
                                                      F02B047-00
     US 6748905
    US 2003164147 A1 US 2002-90500 20020304; WO 2003076783 A2 WO 2003-US5442
ADT
     20030224; AU 2003225593 A1 AU 2003-225593 20030224; US 6748905 B2 US
     2002-90500 20020304
FDT
    AU 2003225593 Al Based on WO 2003076783
PRAI US 2002-90500
                          20020304
IC
         F02B047-00; F02D021-00
     US2003164147 A UPAB: 20040907
     NOVELTY - Engine wear in the operation of an internal combustion engine is
```

h

eb c

g cg b

cg

reduced by recirculating part of exhaust gas from the engine to an intake air supply of the engine; and operating the engine using a water-blended **fuel** composition.

DETAILED DESCRIPTION - Reduction of engine wear in the operation of an internal combustion engine includes recirculating part of exhaust gas from the engine to an intake air supply of the engine; and operating the engine using a water-blended **fuel** composition.

The composition is prepared by combining a normally liquid hydrocarbon **fuel**, water, and surfactant(s).

The surfactant comprises:

- (a) product(s) made from the reaction of an acylating agent with ammonia, an amine, a hydroxyamine, an alcohol, or their mixture of two or more;
- (b) product(s) derived from a polycarboxylic acylating agent, a copolymer derived from olefin monomer(s) and alpha, beta unsaturated carboxylic acid(s) or their derivatives, and a linking compound;
- (c) Mannich reaction product(s) derived from a hydroxy aromatic compound, an aldehyde or a ketone, and an amine containing primary or secondary amino group; and/or
- (d) ionic or nonionic compound(s) having a hydrophilic-lipophilic balance of 1-40.

The linking compound has two or more primary amino groups, primary amino group(s) and secondary amino group(s), at least two hydroxyl groups, or primary or secondary amino groups and hydroxyl group(s).

USE - For reducing engine wear in the operation of an internal combustion engine, e.g. a compression ignition engine.

ADVANTAGE - The invention effectively reduces engine wear in the operation of the internal combustion engine. It reduces the generation of nitrogen oxides (NOx) and particulate emissions in the exhaust of the engine. It achieves engine wear reduction, e.g. piston ring wear reduction or cylinder liner wear reduction.

DESCRIPTION OF DRAWING(S) - The figure is a plot of percent soot in a lubricant versus test hours for engine tests.

Dwg.1/4

FS CPI GMPI

FA AB; GI; DCN

MC CPI: A12-T04C; E05-G09D; E07-D09C; E10-A09B7; E10-A09B8; E10-B01; E10-B03A; E10-C02; E10-D01D; E10-E02D; E10-E02E1; E10-E02F1; E10-G02; E32-A02; H06-D03

L5 ANSWER 6 OF 10 WPIDS COPYRIGHT 2004 THE THOMSON CORP on STN

Full Text

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AN 2004-063955 [07] WPIDS
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CR 2004-224194 [21]; 2004-224195 [21]

DNC C2004-026269

TI Friction modifier e.g. for **fuels**, particularly **gasolines** for internal combustion engines, comprises saturated carboxylic acid salt of alkylated amine.

DC A95 E16 H06

IN ARADI, A A; MALFER, D J; SCHWAB, S D

PA (ETHY) ETHYL CORP; (ETHY) ETHYL INC; (ARAD-I) ARADI A A; (MALF-I) MALFER D J; (SCHW-I) SCHWAB S D

CYC 35

ADT

<u>PI EP 1357170</u> A2 20031029 (200407)* EN 11 C10L001-22

R: AL AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HU IE IT LI LT LU LV MC MK NL PT RO SE SI SK TR

 CA 2424759
 A1 20031024 (200407)
 EN
 C10L001-22

 JP 2004002817
 A 20040108 (200407)
 19 C10L001-22

 US 2003200697
 A1 20031030 (200407)
 C10L001-18

 CN 1453341
 A 20031105 (200408)
 C10L001-10

EP 1357170 A2 EP 2003-7626 20030402; CA 2424759 A1 CA 2003-2424759 20030401; JP 2004002817 A JP 2003-111298 20030416; US 2003200697 A1 US

2002-128529 20020424; CN 1453341 A CN 2003-124017 20030424

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PRAI US 2002-128529
                          20020424
     ICM C10L001-10; C10L001-18; C10L001-22
         C10L001-14; C10L001-32; C10L010-00; C10L010-04; C10M129-32;
          C10M129-40; C10M133-06
          1357170 A UPAB: 20040326
AB
     NOVELTY - A friction modifier comprises saturated carboxylic acid salt of
     alkylated amine.
          DETAILED DESCRIPTION - INDEPENDENT CLAIMS are also included for:
          (a) a fuel additive concentrate comprising 0.2-50 wt.% friction
     modifier comprising branched saturated carboxylic acid salt of mono- or
     di-alkylated amine, 40-99.8 wt.% detergent package comprising detergent
     package comprising detergent and carrier, and 0-80 wt.% solvent;
          (b) a fuel composition comprising a combustible fuel, and 50-2500
     ppm by weight of additive combination;
          (c) a method of preparing a fuel comprising combining a fuel
     which is combustible in gasoline engine having intake valves with a
     fuel additive concentrate; and
          (d) a method of increasing the fuel efficiency in a gasoline
     combustion engine comprising combusting in the engine a gasoline fuel
     comprising a major amount of fuel boiling in the gasoline boiling
     range, and a minor amount of branched saturated carboxylic acid salt of
     alkylated amine.
          USE - The invention is used in fuel additive concentrate for
     fuels (claimed), particularly in gasolines for internal combustion
     engines.
          ADVANTAGE - The invention reduces or eliminates deposits and wear in
     engines, fuel pumps and injectors while imparting enhanced fuel
     economy performance.
     Dwg.0/0
FS
     CPI
     AB; GI; DCN
FA
     CPI: A12-T03B; E05-S; E10-C04L2; H06-B01; H06-D04; H06-D06
MC
     ANSWER 7 OF 10 WPIDS COPYRIGHT 2004 THE THOMSON CORP on STN
L5
   Full
   Text
     2003-457437 [43]
                        WPIDS
ΑN
DNN N2003-363752
                        DNC C2003-121794
     Production of hydrogen gas involves forming water-blended hydrocarbon
ΤI
     feedstock composition comprising hydrocarbon feedstock, water, surfactant
     and water-soluble salt, and steam reforming the blended feedstock.
     E19 E36 H04 H06 L03 X16
DC
     BURRINGTON, J D; GRAHAM, D E; LANGER, D A; MULLAY, J J; YODICE, R
ΙN
     (BURR-I) BURRINGTON J D; (GRAH-I) GRAHAM D E; (LANG-I) LANGER D A;
PA
     (MULL-I) MULLAY J J; (YODI-I) YODICE R; (LUBR) LUBRIZOL CORP
CYC
    99
     WO 2003040030
                     A1 20030515 (200343)* EN
                                                63
                                                      C01B003-32
PI
        RW: AT BE BG CH CY CZ DE DK EE ES FI FR GB GR IE IT LU MC NL PT SE SK
         W: AE AG AL AM AT AU AZ BA BB BG BR BY BZ CA CH CN CO CR CU CZ DE DK
            DM DZ EC EE ES FI GB GD GE GH GM HR HU ID IL IN IS JP KE KG KP KR
            KZ LC LK LR LS LT LU LV MA MD MG MK MN MW MX MZ NO NZ OM PH PL PT
            RO RU SD SE SG SI SK SL TJ TM TN TR TT TZ UA UG UZ VC VN YU ZA ZM
            ZW
     US 2003138373
                     A1 20030724 (200352)
                                                      C01B003-24
                                                      C01B003-32
     EP 1441979
                     A1 20040804 (200451) EN
         R: AL AT BE BG CH CY CZ DE DK EE ES FI FR GB GR IE IT LI LT LU LV MC
            MK NL PT RO SE SI SK TR
     AU 2002343593
                     A1 20030519 (200464)
                                                      C01B003-32
     WO 2003040030 A1 WO 2002-US34917 20021031; US 2003138373 A1 US 2001-8130
ADT
     20011105; EP 1441979 A1 EP 2002-780548 20021031, WO 2002-US34917 20021031;
     AU 2002343593 A1 AU 2002-343593 20021031
```

FDT EP 1441979 A1 Based on WO 2003040030; AU 2002343593 A1 Based on WO

2003040030

PRAI US 2001-8130

20011105

IC ICM C01B003-24; C01B003-32

ICS C01B003-34; C01B003-38; C10L001-32

AB WO2003040030 A UPAB: 20030707

NOVELTY - Hydrogen gas is produced by forming a water-blended hydrocarbon feedstock composition comprising a hydrocarbon feedstock, water, surfactant(s) and water-soluble salt(s); and steam reforming the feedstock to convert the composition to a product comprising hydrogen and carbon oxide(s). The steam is mixed with the feedstock composition to form a vaporized mixture at 50-1200 deg. C.

DETAILED DESCRIPTION - Production of hydrogen gas involves:

- (i) forming a water-blended hydrocarbon feedstock composition comprising a hydrocarbon feedstock, water, at least one surfactant and at least one water-soluble salt; and
- (ii) steam reforming the water blended hydrocarbon feedstock to convert the composition to a product comprising hydrogen and at least one carbon oxide,

The steam is mixed with the feedstock composition to form a vaporized mixture. The temperature of the vaporized mixture is 50-1200 deg. C. The water-blended hydrocarbon feedstock composition is a water-in-oil emulsion, an oil-in-water emulsion or a micro-emulsion. The surfactant comprises:

- (a) a product made from the reaction of an acylating agent with ammonia, an amine, an alcohol or a mixture of at least two of these compounds;
- (b) a product comprising (i) a polycarboxylic acylating agent and (ii) a copolymer derived from olefin monomer(s) and alpha, beta -unsaturated carboxylic acid(s) or their derivative linking group derived from a compound having at least two prim. amino groups, at least two sec. amino groups, at least one prim. amino group and at least one sec. amino group, at least two hydroxyl groups, or at least one prim. or sec. amino group and at least one hydroxyl group;
- (c) an aromatic **Mannich** compound derived from a hydroxy aromatic compound, an aldehyde or a ketone, and an amine containing at least one prim. or sec. amino group;
- (d) an ionic or a nonionic compound having a hydrophilic-lipophilic balance of 1-40; or
 - (e) a mixture of at least two of these four components.
- USE The hydrogen is used for treating a refinery stream or product by hydrocracking, hydrorefining, hydrotreating or hydrodesulfurizing; and for operating a **fuel** cell (both claimed). The hydrogen can also be used in chemical processes including ammonia synthesis from nitrogen (by Haber-Bosch process), aromatic hydrogenation, hydroforming olefinic hydrocarbons to convert them to branched-chain paraffins, preparation of alcohols from synthesis gas, and hydrogenation of fats and oils.

ADVANTAGE - Forming a water-blended hydrocarbon feedstock composition prior to steam reforming improves the efficiency of the steam reforming process and the purity of the hydrogen that is produced. The process provides for lower hydrogen cost due to improved efficiency; lower capital cost or increase in throughput for an existing steam reforming unit due to lower water requirements; lower hydrodesulfurization requirement for the hydrocarbon feedstock; lower carbon monoxide and sulfur impurities for a given condition (level of water and of hydrodesulfurization); and more efficient operation and longer equipment life for **fuel** cell applications e.g. those based on proton exchange membranes. The process allows heavier feedstocks to be handled and transported easily because of their being blended with water.

Dwg.0/0

FS CPI EPI

FA AB; DCN

MC CPI: E11-D; E31-A02; H04-E06; H06-A03; L03-E04F; N02; N07-F EPI: X16-C17

L5 ANSWER 8 OF 10 WPIDS COPYRIGHT 2004 THE THOMSON CORP on STN

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Text
  2002-392730 [42]
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DNC C2002-110420

AN

CR

ΤI

WPIDS

```
automobiles, includes emulsifier comprising amino alkylphenol which is
     made by reacting alkylphenol, aldehyde and amine.
     A95 E19 H06
DC
     FILIPPINI, B B; FORSBERG, J W; MCATEE, R J; MORETON, D J; STECKEL, T F
IN
     (FILI-I) FILIPPINI B B; (FORS-I) FORSBERG J W; (MCAT-I) MCATEE R J;
PA
     (MORE-I) MORETON D J; (STEC-I) STECKEL T F; (LUBR) LUBRIZOL CORP
     97
CYC
                     A1 20020221 (200242)*
     US 2002020106
                                                14
                                                       C10L001-32
PI
     WO 2003002693
                     A2 20030109 (200305)
                                                       C10L001-00
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            KZ LC LK LR LS LT LU LV MA MD MG MK MN MW MX MZ NO NZ OM PH PL PT
            RO RU SD SE SG SI SK SL TJ TM TN TR TT TZ UA UG UZ VN YU ZA ZM ZW
                     A1 20030303 (200452)
                                                       C10L001-00
     AU 2002303992
    US 2002020106 A1 CIP of US 1999-349268 19990707, CIP of US 1999-390925
ADT
     19990907, CIP of US 2000-483481 20000114, CIP of US 2001-755577 20010105,
     US 2001-892073 20010626; WO 2003002693 A2 WO 2002-US18405 20020611; AU
     2002303992 A1 AU 2002-303992 20020611
    AU 2002303992 Al Based on WO 2003002693
FDT
PRAI <u>US 2001-892073</u>
                          20010626; US 1999-349268
                                                          19990707;
                          19990907; US 2000-483481
     US 1999-390925
                                                          20000114;
     US 2001-755577
                          20010105
IC
     ICM C10L001-00; C10L001-32
AΒ
     US2002020106 A UPAB: 20040813
     NOVELTY - An aqueous hydrocarbon fuel emulsion comprises water, fuel
     and an emulsifier. The emulsifier comprises an amino alkylphenol, which is
     made by reacting alkylphenol, an aldehyde and an amine.
          DETAILED DESCRIPTION - An INDEPENDENT CLAIM is also included for a
     process for making an aqueous hydrocarbon fuel by mixing a liquid
     hydrocarbon fuel and at least one emulsifier to form a hydrocarbon
     fuel emulsifier mixture, and mixing the hydrocarbon fuel emulsifier
     mixture with water or water and ammonium nitrate under emulsification
     conditions to form the aqueous hydrocarbon fuel composition. The aqueous
     hydrocarbon fuel composition has a discontinuous phase having aqueous
     droplets of at most 1 mu m mean diameter.
          USE - For use in fueling an engine used in, e.g. automobiles, trucks
     or urban buses.
          ADVANTAGE - The inventive water in fuel emulsion is stable in
     storage. The emulsifier provides good stability for water emulsified fuel.
     Dwg.0/0
FS
    CPI
FΑ
    AB; GI; DCN
     CPI: A10-E01; A12-T03B; A12-W12C; E06-H; E07-A02C; E07-E03; E07-H;
MC
          E10-A03; E10-A05; E10-A16B; E10-A22; E10-A25A2; E10-A25B2; E10-B01;
          E10-B03A; E10-B03B; E10-B03B1; E10-B04; E10-C02; E10-D01D; E10-E02U;
          E10-G03; E32-A02; E32-A04; E33; E34; H06-B
    ANSWER 9 OF 10 WPIDS COPYRIGHT 2004 THE THOMSON CORP on STN
L5
   Full
   Text
ΑN
     1991-030962 [05]
                        WPIDS
```

New polymeric amino- or thio-alkylene O-substd. phenol(s) - have substd.

aldehyde residues with para residue linked to poly amine, used as liq.

2000-283277 [24]; 2001-159404 [16]; <u>2001-588949</u> [66]; <u>2002-739803</u> [80]

Aqueous hydrocarbon fuel emulsion for fueling engine used in, e.g.

eb

A95 E13 E14 H06

hydrocarbon antioxidants.

BAGWELL, T V; WEERS, J J

C1991-013222

DNC

ΤI

DC

ΙN

```
PA
     (PETL) PETROLITE CORP
CYC
     EP 410577
                      A 19910130 (199105)*
PΙ
         R: BE DE ES FR GB IT NL
     CA 2019320
                      Α
                         19910127 (199116)
                         19910910 (199139)
     US 5047069
                      Α
     EP 410577 A EP 1990-306874 19900622; US 5047069 A US 1990-601151 19901022
ADT
                            19890727
PRAI US 1989-386337
     US 3269810; US 3877889; US 3935160; US 4217235
REP
     C07C215-30; C07D295-12; C08G008-38; C08G012-46; C08G014-14; C10L001-22;
     C10L005-00
            410577 A UPAB: 19930928
AB
     Cpd. of formula (I): where X=N or S; R=H or a 1-5C alkyl; R1, R2=5-18C
     alkyl or cycloalkyl which may opt. contain a heteroatom, only one of R1
     and R2 being present if X=S, or may be combined to form a ring; R6=residue
     of 1-6C aldehyde; R7=residue of a polyamine of formula (II) where R3=8-50C
     alkyl, opt. contq. ether linkages; R4, R5=2-6C alkylene; a=0 or 1; and
     m=at least 3. Pref. prepg. comprises polymeric reaction of a phenol, or
              with -CHR-XR1R2; a polyamine of formula (II); and a 1-6C aldehyde
     claimed. A Mannich base reaction may be carried out by various methods.
     Pref. aldehyde=formaldehyde.
          USE/ADVANTAGE - Antioxidant for liq. HCs (as 0.05-500 ppm) in
     compsns. contq. fuel, kerosene or diesel fuel (claimed). The cpds.
     prevent discolouration and sludge formation and sequester metals which can
     cause degradation.
     0/0
FS
     CPI
FΆ
     AB; DCN
MC
     CPI: A05-B01; A12-T03A; E07-H; E10-B01D; H06-D01
L5
     ANSWER 10 OF 10 WPIDS COPYRIGHT 2004 THE THOMSON CORP on STN
   Full
   Text
ΑN
     1989-250597 [35]
                         WPIDS
DNC
     C1989-111640
     Oil-soluble additive package - comprising lubricating oil ashless
ΤI
     dispersants or viscosity index improver dispersant, demulsifier.
DC
     A97 E19 H08 H09
     COLCORD, L J; EMERT, J; MERCHANT, P; PACANSKY, T J; WADDOUPS, M
IN
PΑ
     (ESSO) EXXON CHEM PATENTS INC
CYC
     18
                      A 19890830 (198935) * EN
                                                   40
     EP 330522
PΙ
         R: AT BE CH DE ES FR GB IT LI NL SE
     AU 8930710
                      A 19890831 (198943)
     NO 8900785
                      A 19890918 (198943)
     DK 8900880
                      Α
                        19890827 (198944)
     BR 8900859
                      A ·19891017 (198947)
                      A 19900912 (199043)
     JP 02229892
                                                         C10M149-22
     IL 89210
                      Α
                        19920621 (199234)
     CA 1331599
                         19940823 (199435)
                                                          C10M161-00
                      C
                                                   47
     EP 330522
                      B1 19941012 (199439)
                                              EN
                                                         C10M161-00
         R: AT BE CH DE ES FR GB IT LI NL SE
     DE 68918728
                      Ε
                         19941117 (199445)
                                                         C10M161-00
                                                          C10M161-00
     ES 2060752
                      T3 19941201 (199504)
                                                   33
     JP 2696380
                      B2 19980114 (199807)
                                                         C10M161-00
     EP 330522 A EP 1989-301931 19890227; JP 02229892 A JP 1989-43363 19890227;
ADT
     IL 89210 A IL 1989-89210 19890207; CA 1331599 C CA 1989-590597 19890209;
     EP 330522 B1 EP 1989-301931 19890227; DE 68918728 E DE 1989-618728
     19890227, EP 1989-301931 19890227; ES 2060752 T3 EP 1989-301931 19890227;
     <u>JP 2696380 B2 JP 1989-43363</u> 19890227
     \underline{\text{DE }68918728} E Based on \underline{\text{EP }330522}; \underline{\text{ES }2060752} T3 Based on \underline{\text{EP }330522}; \underline{\text{JP }}
FDT
     2696380 B2 Previous Publ. <u>JP 02229892</u>
PRAI US 1988-160690
                           19880226
     A3...9006; <u>DE 1794133</u>; <u>EP 32617</u>; <u>EP 74618</u>; No-SR.Pub; <u>US 2996551</u>; US
REP
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3429817; <u>US 3509052; US 3511882</u>
     ICM C10M149-22; C10M161-00
IC
         C10L001-18; C10L001-22; C10L001-24; C10M129-00; C10M133-52;
          C10M135-10; C10M143-00; C10M145-02; C10M145-16; C10M145-26;
          C10M149-12; C10M157-00; C10M157-04; C10M159-16; C10M165-00;
          C10M167-00; C10N030-02
    C10N030:00, C10N030:02, C10N030:04, C10N040:25; C10M129:00, C10M129:95,
          C10M133:52, C10M135:10, C10M135:24, C10M143:00, C10M145:02,
          C10M145:04, C10M145:06, C10M145:26, C10M149:00, C10M159:16,
          C10M161-00; C10M143:00, C10M145:02, C10M145:26, C10M145:34,
          C10M145:36, C10M149:00, C10M165-00; C10M129:18, C10M133:52,
          C10M135:24, C10M143:00, C10M145:02, C10M145:06, C10M145:26,
          C10M149:00, C10M159:16, C10M161-
AB
     ΕP
           330522 A UPAB: 19930923
     ashless additive comprising at least one of (1) ashless dispersants
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Oil-soluble mixt useful as an oil additive comprises (A) a lubricating oil ashless additive comprising at least one of (1) ashless dispersants comprising (i) oil-soluble salts, amides, imides, oxazolines and/or esters of long chain hydrocarbon substd mono- and dicarboxylic acids or their anhydrides; (ii) long chain aliphatic hydrocarbon having a directly attached polyamine; and (iii) Mannich condensation prods of 1 mole of a long chain hydrocarbon substd phenol with 1-2.5 moles HCHO and 0.5-2 moles polyalkylene polyamine; where the long chain hydrocarbon gps above are polymers of 2-10C, eg 2-5C monoolefins, and have a no ave mol wt of at least 900.

(2) polymeric viscosity index improver dispersants including (i) polymers comprised of 4-24C unsatd esters of vinyl alcohol or 3-10C unsatd mono- or dicarboxylic acid with unsatd 4-20C N-contg monomers; (ii) polymers of 2-20C olefin with unsatd 3-10C mono- or dicarboxylic acid neutralised with amine, hydroxyamine or alcohol; and (iii) polymers of ethylene with a 3-20C olefin further reacted either by grafting 4-20C unsatd N-contg monomers or by grafting an unsatd acid onto the backbone and then reacting the carboxylic acid gps with amine, hydroxyamine or alcohol; or (3) mixts of (1) and (2); and (B) a demulsifier additive comprising the reaction prod of alkylane oxide and an adduct obtd by reacting a bisepoxide with a polyhydric alcohol.

USE/ADVANTAGE - Useful in **fuels**, eg middle distillates, in lubricating oil compsns used as crankcase lubricants for automobile, marine and railway engines; in power transmitting fluids, eg automatic transmission fluids, tractor fluids, hydraulic fluids, power steering fluids, etc; in gear lubricants, industrial oils, pump oils etc. Tendency to **emulsion** formation, esp in moist or humid conditions is reduced. storage stability of the package is increased by use of the compatibility additive.

0/0 CPI

FA AB

FS

MC CPI: A12-T03B; A12-W02A; E07-E01; E07-H03; E10-A24; E10-A25; E10-B01A2; E10-B01A4; E10-B01E; E10-D03; E10-E04J; E10-G02G; E10-G02H; H06-D; H07-G08; H08-D05

=> file stnguide SINCE FILE TOTAL COST IN U.S. DOLLARS ENTRY SESSION 100.71 100.92 FULL ESTIMATED COST DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS) SINCE FILE TOTAL ENTRY SESSION CA SUBSCRIBER PRICE -5.60 -5.60

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AND TECHNOLOGY CORPORATION, AND FACHINFORMATIONSZENTRUM KARLSRUHE

FILE CONTAINS CURRENT INFORMATION.
LAST RELOADED: Dec 10, 2004 (20041210/UP).

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